

GOLIK, A.Z. [Holyk, O.Z.]; SHIMANSKIY, Yu.I. [Shymans'kyi, IU.I.]; KOBIYCHUK, N.M.  
[Kobiichuk, N.M.]

Compressibility of isoviscous substances [with summary in English].  
Ukr.fiz.zhur. 3 no.4:537-541 J1-Ag '58. (MIRA 11:12)

1. Kiyevskiy gosudarstvennyy universitet.  
(Compressibility)

GOLIK, A.Z.; KOCHARNYUK, R.F.

Physical properties and structure of normal alcohol solutions  
in acetone. Ukr. khim. zhur. 24 no.1:29-32 '58. (NIRA 11:4)

1.Kiyevskiy gosudarstvennyy universitet im. T.G. Shevchenko.  
(Alcohols) (Acetone) (Solution (Chemistry))

KOTORLENKO, L.A.; GOLIK, A.Z.; KOVNERISTAYA, A.S.

Viscosity and electric conductivity of lithium chloride solutions in  
alcohols. Ukr.khim.zhur. 24 no.5:618-625 ' 58. (MIRA 12:1)

1. Kiyevskiy gosudarstvennyy universitet imeni T.G. Shevchenko.  
(Lithium chloride) (Solution (Chemistry))

GOLIK, A.Z.; SOLOMKO, V.P.

Investigation of the physical properties of the water - acetone -  
alcohol system. Part 1: Water - acetone-ethanol system. Ukr.khim.zhur.  
24 no.6:734-740 '58.  
(MIRA 12:3)

1. Kiyevskiy gosudarstvennyy universitet im. T.G. Shevchenko.  
(Acetone) (Ethyl alcohol) (Systems (Chemistry))

GOLIK, A.Z. [Holyk, O.Z.

Viscosity and electrical conductivity of zinc and cadmium  
amalgams. Part 2. Ukr.fiz.zhur. 4 no.4:421-426 J1-Ag '59.  
(MIRA 13:4)

1. Kiyevskiy gosudarstvennyy universitet im. P.G. Shevchenko.  
(Amalgams) (Zinc) (Cadmium)

GOLIK, A.Z. [Holyk, O.Z.]; SHYMANSKAYA, Ye.T. [Shymanska, O.T.]

Investigation of the critical state of substances by Isoperic method. Part 2. Temperature dependence of the density of hexane near the critical point. Ukr.fiz.zhur. 4 no.6:769-788 M-D 159.  
(MIRA 24:10)

1. Kiyevskiy gosudarstvennyy universitet im. N.G.Shevchenko.  
(Hexane--Thermal properties)

GOLIK, A.Z.; SOLOMKO, V.P.

Investigation of the physical properties of the system water-acetone-alcohols. Part 2: System water-acetone-butanol. Ukr. khim. zhur. 25 no.1:40-44 '59. (MIRA 12:4)

1. Kiyevskiy gosudarstvennyy universitet in. T.G. Shevchenko.  
(Water) (Acetone) (Butyl alcohol)

RUSSIAN BASIC INFORMATION 307/5469

Sovetskaniye po kriticheskim yavleniyam i flyuktuatsiyam v rastvorakh. Moscow, 1960.

Kriticheskiye yavleniya i flyuktuatsii v rastvorakh; teziy i doklady, yanvar' 1960 g. (Critical Phenomena and Fluctuations in Solution; Transactions of the Conference, January 1960) Moscow, Izd-vo AN SSSR, 1960. 190 p. 2,900 copies printed.

Sponsoring Agencies: Akademiya nauk SSSR. Otdeleniye khimicheskikh nauk. Moskivskiy gosudarstvennyy universitet im. M. V. Lomonosova. Khimicheskii fakul'tet.

Responsible Ed.: M. I. Shokhgarenov, Doctor of Chemical Sciences, Professor; Ed. of Publishing House: E. S. Dragunov; Tech. Ed.: S. G. Tikhomirova.

PURPOSE: This collection of articles is intended for scientific personnel concerned with chemistry, physics, and heat power engineering.

Card 1/9



Critical Phenomena and Fluctuations

SOV/5-83

CONTENTS. The book contains 21 of the 22 reports read at the Conference on Critical Phenomena and Fluctuations in Leningrad, organized by the Chemical Division of Russian State University, January 20-23, 1980. The reports contain results of investigations carried out in recent years by Soviet scientists in the field of critical phenomena. The Organizing Committee of the Conference was composed of Professor Kh. I. Molodtsov, A. A. Golik, I. R. Kiselevsky (Chairman), V. K. Serebrenko, A. V. Storonkin, I. A. Fisher, and M. I. Shalcharenov (Deputy Chairman). References accompany individual articles.

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Critical Phenomena and Fluctuations

Gov/5859

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Andrey, M. V., B. I. Kolyanov, and N. G. Spiridovich [Department of Physics, Belarusian State University].  
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Critical Phenomena and Fluctuations

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Smirnov, V. P., and Yu. D. Kolpakov [Laboratory of Molecular Physics, Ural Polytechnic Institute imeni S. M. Kirov, and the Laboratoriya teplofiziki, Ural'skiy filial AN SSSR -- Thermophysics Laboratory, Ural Branch, AS USSR]. Light Scattering in Carbon Dioxide along Pre- and Post-Critical Isotherms

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Smirnov, B. A. [Institut neftekhimicheskogo sinteza AN SSSR -- Card 7/9

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Critical Phenomena and Fluctuations

1977/1/69

Shchegoleva, Ye. T., Ye. I. Shchegolev, and A. E. Golia (Laboratory of Molecular Physics, Division of Physics, Moscow State University named G. G. Shchegolev). Investigation of the Critical State of Pure Substances by Taylor's Method

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10-10-01

Card 9/9

15.8500

25577  
S/185/60/005/002/012/022  
D274/D304

AUTHORS: Golyk, O.Z. and Cholpan, P.P.  
TITLE: Molecular structure, compressibility, surface tension and viscosity of certain polysiloxanes  
PERIODICAL: Ukrayins'kyy fizychnyy zhurnal, v. 5, no. 2, 1960, 242-250

TEXT: Polymethyl- and polyethylsiloxanes with linear molecules are experimentally studied, this article being a continuation of one of the authors previous works: O.Z. Golyk (Ref. 2: UkhZh, 23, no. 2, 139, 1957, and 2 articles in collaboration with others). From intensity curves of X-ray scattering, electron-density curves were constructed; these were used for determining the valence angles, the length of the chemical bond, and the packing of the molecules in the liquid state. The intensity curves, plotted on figures, show that polymethyl- and polyethylsiloxanes with linear molecules have a similar structure in the liquid state. The density, surface tension, compressibility and viscosity of these substances were

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D274/D304

Molecular structure...

investigated for a wide temperature range; figures and tables are given with the results of these investigations. For polymethylsiloxanes, the polytherms of surface tension and of viscosity are the higher, and those of compressibility - the lower, the higher the potential of intermolecular forces, and the higher the critical temperature of the substance. The surface tension is also in direct proportion with the size of the molecules. Adiabatic compressibility of polymethylsiloxanes was studied by means of an ultrasonic interferometer. The temperature dependence of viscosity follows an exponential law. The polytherms of surface tension and of viscosity in the case of polyethylsiloxanes, are also the higher, the higher the potential of intermolecular forces and the higher the critical temperature. The activation energy too, is in direct proportion with intermolecular potential and critical temperature. The viscosity of binary solutions of polymethylsiloxanes was also studied, and isoviscous substances were obtained; both the activation energy and also compressibility of the isoviscous substances is practically the same. This study gives additional proof of the correspondence between structure and intermolecular forces on the

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D274/D304

Molecular structure...

one hand, and surface tension, compressibility, and viscosity on the other. There are 9 figures, 4 tables and 3 references: 2 Soviet-bloc and 1 non-Soviet-bloc. The reference to the English-language publication reads as follows: H.S. Green, The molecular theory of fluids, Amsterdam, 1952; I.J. Kirkwood a. F.P. Buff, J. Chem. Phys., 17, 338, 1949; I.J. Kirkwood, F. P. Buff, H.S. Green, J. Chem. Phys., 17, 998, 1949.

ASSOCIATION: Kyyvs'kogo ordena Lenina universytetu im. T.G. Shevchenka (Kiyev Order of Lenin University im. T.G. Shevchenko), Department of Molecular Physics

SUBMITTED: October 1, 1959

Card 3/3

GOLIK, A.Z. [Holiyk, O.Z.]; CHOLPAN, P.F. [Cholpan, P.P.]

Molecular structure and physical properties of some polysiloxanes. Part 2: Structure and physical properties of isoviscous polysiloxanes. Ukr. fiz. zhur. 5 no.6:843-849 N-D '60.  
(MIRA 14:3)

1. Kiyevskiy ordena Lenina gosudarstvennyy universitet im. I.G. Shevchenko.

(Siloxanes)

GOLIK, A.Z. [Holyk, O.Z.]; CHOLPAN, P.F. [Cholpan, P.P.]

Molecular structure and physical properties of some polysiloxanes.  
Part 3: Viscosity, compressibility, and structure of liquid cyclic  
polysiloxanes. Ukr. fiz. zhur. 5 no.6:850-856 N-D '60.

(MIRA 14:3)

1. Kiyevskiy ordena Lenina gosudarstvennyy universitet im. T. G.  
Shevchenko.

(Siloxanes)

SKRYSHEVSKIY, Anton Frantsevich; GOLIK, A.Z., prof., otv. red.;  
DROZHNIN, E.V., red.; OKOITAYA, Ye.D., tekhn. red.

[Diffraction of X rays, electrons, and neutrons in gases and  
the molecular structure] Difraktsiya rentgenovskikh luchei,  
elektronov i neitronov v gazakh i stroenie molekul. Kiev, Izd-  
vo Kievskogo univ., 1961. 84 p. (MIRA 15:9)  
(X rays--Diffraction) (Electron diffraction examination)  
(Neutrons--Diffraction)

GOLIK, A.Z.; CHOLPAN, P.F.

Speed of ultrasound in some polysiloxanes. Akust.zhur. 7 no.1:33-39  
:61. (NIRA 14:4)

1. Kiyevskiy gosudarstvennyy universitet.  
(Siloxanes)  
(Ultrasonic waves)

5/68: 6 /000/021700/C94  
E/C2/E158

AUTHORS: Shimanskaya Ye. T., Shimanskaya Ye. I., Golik, A. N.

TITLE: Investigation of the critical state of pure substances by  
Teppler's method

PERIODICAL: Referativnyy zhurnal. Khimiya no. 2 1961. 43, abstract  
21B347 (Sb. "Kritich. yavleniya i fluktuatsii v rastvorakh";  
M., AN SSSR, 1960 171 188)

TEXT: A method has been developed for the investigation of critical  
states, by means of which the density  $\rho$  of a substance can be measured  
in any point in a chamber (by the optical Teppler method) with long-time  
thermostating. The apparatus is described in detail. Heptane and hexane  
were examined. Density has a non-monotonic gradient with respect to the  
chamber height  $Z$  and has a maximum at the meniscus. This maximum  
increases as the temperature approaches the point  $T_m$  at which the meniscus  
vanishes. With a steady temperature change rate, e.g., 1 deg/hr, the  $\rho(Z)$   
maximum is present on heating and absent on cooling (i.e., a hysteresis  
is observed). With irregular changes in temperature and long-time  
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S/OB/11,000/02 /010/034  
B-02/B-16

Investigation of the critical

(15 20 hr) thermodynamic, the  $dp/dZ$  maximum is however present on heating as well as cooling, the maxima are then lower than in the case of steady heating. The  $dp/dZ$  curves are found by integrating  $dp/dZ = f(Z)$ . For  $T > T_m$  they are S-shaped. In the lower part of the chamber density is higher, and in the upper part lower than critical. This is in full agreement with classical representations regarding the existence of a critical point and not a region, even allowing for the effect of gravitational field. The critical state is realized only in a narrow layer at the point where the meniscus vanishes. Above and below this layer the substance is not in a critical state although its temperature is critical. As the density difference throughout the chamber corresponds at the critical temperature to the equilibrium state then it must be assumed that displacement sometimes occurs, levelling the density and removing the system from the state of equilibrium. [Abstracter's note: Complete translation.]

Card 2/2



GOLIK, A.E.; KARANOVSKIY, V.Ye.

Heat of vaporization, composition of vapors, and surface tension  
of solutions of paraffins and alcohols. Ukr.khim.zhur. 27 no.5:  
574-577 '61. (MIRA 14:9)

1. Kiyevskiy gosudarstvennyy universitet im. T.G. Shevchenko.  
(Paraffins) (Alcohols)

YOLIN, A.S.; BUCHOVSKIY, V.Ye.

Latent heat of vaporization of alcohols in acetone solutions.  
Ukrainian. 27 no.5:577-580 '61. (MIRA 14:9)

1. Kiyevskiy gosudarstvennyy universitet im. T.G. Shevchenko.  
(Alcohols) (Heat of vaporization)

15 8:70

S/073/6-1017/10001/10  
B110/B117

AUTHORS: Golik, A. Z., Cholpan, P. F., Ivanova, I. I.

TITLE: Investigation of some physical properties of methylphenyl siloxanes

PERIODICAL: Ukrainskiy khimicheskiv zhurnal, v. 17, no. 5, 1967, pp. 754 - 759

TEXT: This work is an investigation of viscosity, surface tension and adiabatic compressibility of 1,5-dimethylphenyl-3-methylphenylsiloxane  $(CH_3)_2C_6H_4SiOSi(C_6H_4CH_3)OSiC_6H_4(CH_3)_2$ ; 1,5-trimethylphenyl-3-methylphenyltrisiloxane  $(CH_3)_3SiOSi(C_6H_4CH_3)OSi(CH_3)_3$ ; 1,5-trimethylphenyl-3,5-dimethylphenyltetrasiloxane  $(CH_3)_3Si[OSiCH_2C_6H_4]_2OSi(CH_3)_3$ ; polymer 1 (P1)  $(CH_3)_3Si[OSiCH_2C_6H_4]_3OSi(CH_3)_3$ ; polymer 2 (P2)  $(CH_3)_3Si[OSiCH_2C_6H_4]_4OSi(CH_3)_3$ ; polymer 3 (P3)  $(CH_3)_3Si[OSi(C_6H_4)]_3OSi(CH_3)_3$ .

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3/21/77  
E11/1117

Investigation of some physical...

$-\text{OSi}(\text{CH}_3)_3$ ; polymer 1 (P4)  $(\text{CH}_3)_3\text{Si}[\text{OSi}(\text{CH}_3)_2]_n\text{OSi}(\text{CH}_3)_3$

structure of the polymethyl polysiloxane was determined and their molecular weight was determined. The viscosity of the polymers with temperature according to the Arrhenius law. The polymers of the viscosities of P1 and P2 were determined. Only the first three substances correspond to the formula.

$\eta \cdot C/(V \cdot d)$  (Table 1) Between 10 and 100°C, the viscosity is increasing temperature. Ultrasonic speed was measured by interferometer by I. G. Mikhailevich. The length of ultrasonic waves,  $\lambda$ , generated by a generator of the trimer with 4  $\text{C}_2\text{H}_5$  groups (A), of the tetramer with 2  $\text{C}_2\text{H}_5$  groups (B), of the methyl trimer (C), and of the tetramer (D). The length of ultrasonic waves,  $\lambda$ , generated by a generator of the tetramer with two  $\text{C}_2\text{H}_5$  groups (E) and of the trimer with one  $\text{C}_2\text{H}_5$  group (F) were compared with the polymers of A, B, C, D, E, F.

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3/073/61/027/006/002/005

B110/B147

Investigation of some physical...

polytherm of D lies above that of E. In P1, P2, and P3, a slight deviation from linearity was found at 40°C (near their solidification point). Adiabatic compressibility was calculated by;  $\beta = 1/a^2$  ( $a$  = ultrasonic speed,  $\rho$  = density,  $\beta$  = adiabatic compressibility. It is inversely proportional to the number of phenyl radicals. From the linear dependence;  $\ln \beta = f(t)$ ,  $\beta = \beta_0 \exp(T/C)$  is derived;  $T$  = experimental temperature,  $\beta_0$  = adiabatic compressibility at  $T = 0$ ,  $C$  = constant (Table 2). There are 10 figures, 2 tables, and 3 Soviet references.

ASSOCIATION: Kiyevskiy gosudarstvennyy universitet im. T. G. Shevchenko  
(Kiev State University imeni T. G. Shevchenko)

SUBMITTED: September 29, 1960

Card 3/5

GOLIK, A.Z.; KLASSEN, I.F.; KUCHAK, G.M.

Speed of propagation of ultrasonic waves in certain zinc and cadmium  
amalgams. Akust.zhur. 7 no.2:258-260 '61. (MDRA 14:7)

1. Kiyevskiy gosudarstvennyy universitet.  
(Ultrasonic waves--Speed) (Zinc amalgam)  
(Cadmium amalgam)

GOLIK, A.Z., ref., etv. red.; ROSECHINA, G.P., dots., etv. red.;  
MIRONETS, Ye.M., red.; KHECHANOVSKAYA, T.I., tekhn. red.

[Structure and physical properties of matter in the liquid  
state; materials] Stroenie i fizicheskie svoystva veshche-  
stva v zhidkom sostoianii; materialy. Kiev, Izd-vo Kievskogo  
univ., 1962. 146 p. (MIRA 15:6)

1. Soveshcheniye posvyashchennoye probleme zhidkogo sostoyaniya  
veshchestva. 4th, Kiev, 1950.

(Liquids)

GOLIK H.2.

STRUCTURE AND PHYSICAL PROPERTIES OF MATTER IN A LIQUID STATE  
reports read at the 4th Conference convened in KIEV from 1 to 5 June  
1969, published by the Publisher House of KIEV University, KIEV,  
USSR, 1962

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GOLIK, A.Z.

STRUCTURE AND PHYSICAL PROPERTIES OF MATTER IN A LIQUID STATE  
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1969, published by the Publishing House of KIEV University, KIEV,  
USSR, 1969

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ROSHCHINA, Galina Petrovna; GOLIK, A.Z., prof., otv. red.; VIKO,  
V.I., red.; ~~... ..~~, ~~...~~, ~~...~~ red.

[Molecular scattering of light in gases] (Molekularnoe ras-  
seyaniye sveta v gazakh. Kiev, Izd-vo Kievskogo univ., 1971,  
37 p.) (MIRA 15:11)  
(Gases--Optic) (Scattering (Optical)) (Light--Scattering)

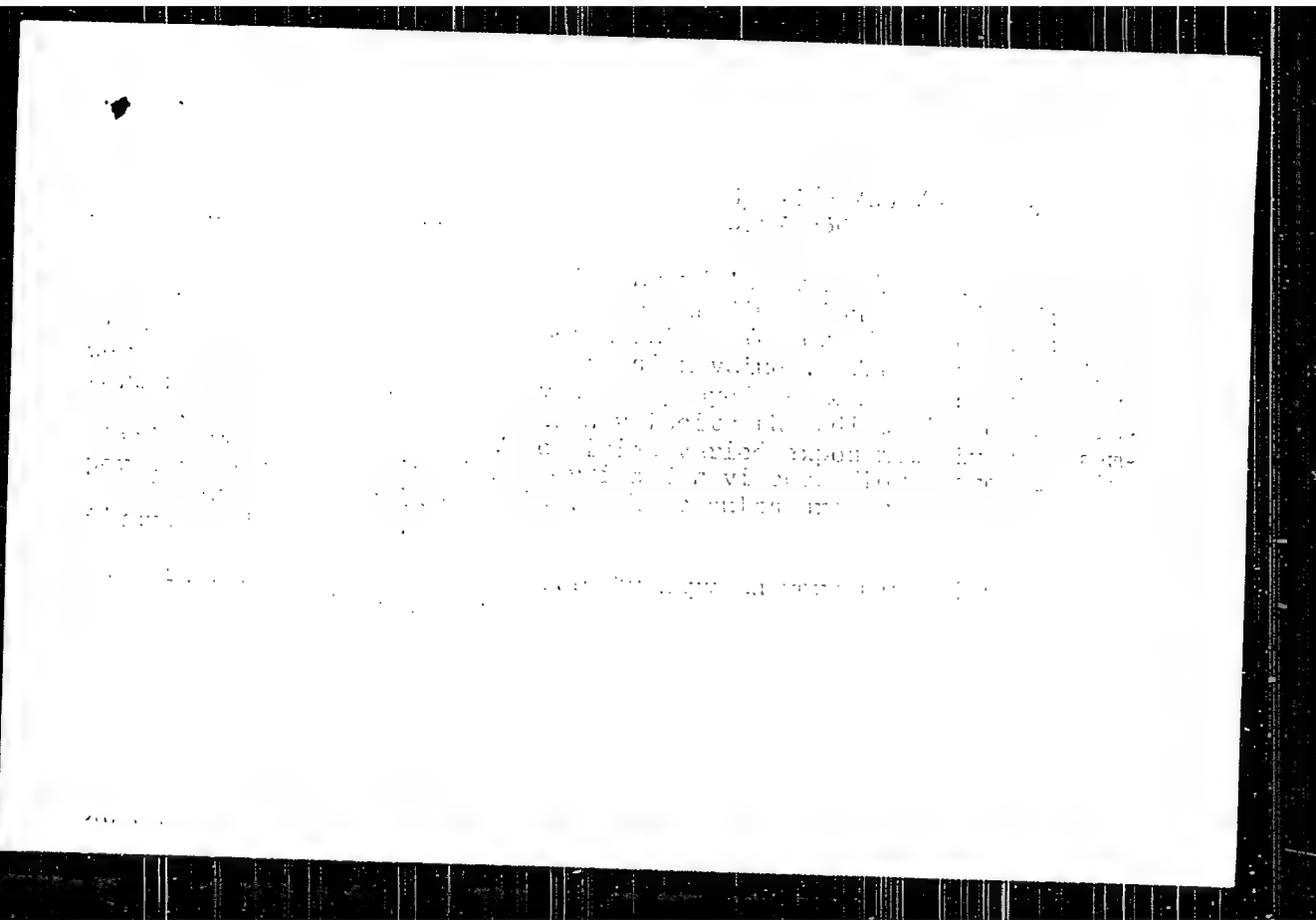
GOLIK, A.Z., prof., otv. red.; LOSHCINA, G.F., doc., otv. red.;  
- MIRONETS, Ye.M., red.; KHOKHANOVSKAYA, T.I., tekhn.red.

[Structure and physical properties of matter in the liquid  
state; materials] Stroenie i fizicheskie svoistva veshchestva  
v zhidkom sostoianii; materialy. Kiev, Izd-vo Kievskogo univ.,  
1962. 146 p. (MIRA 15:9)

1. Soveshchaniye posvyashchennoye probleme zhidkogo sostoyaniya  
veshchestva. 4th, Kiev, 1959.

(Liquids)





U/643/62/000/000/006/010  
J207/0308

AUTHOR: Golik, A.M. and Klassen, I.F.

TITLE: Relationship of the viscosity and electrical conductivity with the structure of zinc and cadmium amalgams

ABSTRACT: Ispytaniye i fizicheskiye svoystva veshchestva v zhidkom sostoyanii; materialy IV soveshch. po probl. zhidkogo sost. veshchestva, v Kiyevе 1959 g. Kiev, Izd-vo Kiev. univ., 1962, 96-100

NOTE: The purpose of this work was to check the hypothesis that both the first (shear) viscosity and the electrical conductivity of liquid metals and their solutions are related to the short-range order. The viscosity, density and electrical conductivity of zinc and cadmium amalgams were measured at temperatures up to 350°C in a wide range of compositions. Amalgams with the same viscosity had the same short-range order but different electrical conductivities. Amalgams with the same electrical conductivity had practically the

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Relationship of the viscosity ...

J/845/62/000/000/006/010  
D207/D308

same density but different viscosities. The results confirm the hypothesis cited above. There are 4 figures and 1 table.

ASSOCIATION: Kiyevskiy gosudarstvennyy universitet (Kiev State University)

Card 2/2

GOLIK, A.Z.; RYNDICH, N.A.; KUCHINKA, M.Yu.; ANDRIYENKO, S.S.

Thermomechanical properties of cord made from polycaprolactam.  
Khim.volok. no.2:23-25 '62. (MIRA 15:4)

1. Kiyevskiy gosudarstvennyy universitet im. Shevchenko.  
(Textile fibers, Synthetic) (Azepinone)



GOLIK, A.Z., [Holyk, O.Z.]; CHOLPAN, P.F. [Cholpan, P.P.]

Molecular structure and physical properties of certain siloxanes.  
Part 4. Density of two-component solutions of liquid siloxanes.  
Ukr.fiz.zhur. 7 no.5:549-553 My '62. (MIRA 16:1)

1. Kiyevskiy gosudarstvennyy universitet im. Shevchenko.  
(Siloxanes)

GOLIK, A.Z. [Holyk, O.Z.]; CHOLPAN, P.F. [Cholpan, P.P.]

Molecular structure and physical properties of certain siloxanes.  
Part 5. Surface tension and molecular interaction of liquid  
siloxanes. Ukr.fiz.zhur. 7 no.5:554-558 My '62. (MIRA 16:1)

1. Kiyevskiy gosudarstvennyy universitet im. Shevchenko.  
(Siloxanes)

GOLIK, A.Z. [Holyk, O.Z.]; CHOLPAN, P.F. [Cholpan, P.P.]

Density and short-range coordination of certain liquids. Ukr.  
fiz.zhur. 7 no.5:559-562 My '62. (MIRA 16:1)

1. Kiyevskiy gosudarstvennyy universitet im. Shevchenko.  
(Liquids)

GOLIK, A.Z.

Present stage and problems of the physics of liquids. Ukr.fiz.  
zhur. 7 no.7:685-686 J1 '62. (MIRA 15:12)  
(Liquids)

S/185/62/007/008/001/008  
D234/D303

3.7

AUTHOR: Golik, A.Z.

TITLE: Connection of compressibility and sheer viscosity  
with the structure of liquid state of matter

PERIODICAL: Ukrayins'kyy fizychnyy zhurnal, v. 7, no. 8, 1962,  
806 - 811

TEXT: The author gives the experimental values of compressibilities and shear viscosities of several paraffins, methylsiloxanes and alcohols, obtained by him in collaboration with P.F. Cholpan and I.I. Ivanova. Graphs of temperature dependence of these quantities are given for all substances mentioned. The compressibility was determined from data on density and ultrasound velocity, the latter being measured by I.T. Mikhaylov's interferometer. The empirical formula for the temperature dependence of the compressibility  $\beta = \beta_0 \exp [\alpha T]$  is found to agree with experiment better than the expression given by the cell theory. The temperature dependence of viscosity is described by Ya.I. Frenkel's formula  $\eta = A \exp$   
Card 1/2

Connection of compressibility and ...

S/185/52/007/008/001/008  
D234/D308

[B/RT]. A table of critical temperatures and the values of  $\rho_0$ ,  $\alpha$ ,  $\beta$  and  $A$  is included. It is found that there is a correlation between  $B$  and  $\alpha$ . Properties of isoviscous substances for each group were studied. The author gives as an example the graph of intensity of scattered X-rays, plotted against the scattering angle, for  $(CH_3)_{10}Si_4O_3$  and the solution consisting of 48.73% of  $(CH_3)_8Si_3O_2$  and 51.27% of  $(CH_3)_{12}Si_3O_4$ , isoviscous with the former. All experimental points are situated on the same curve, which indicates that the isoviscous substances have the same structure. There is 1 table and 8 figures.

ASSOCIATION: Kiyevskiy universitet (Kiev University)

Card 2/2

1. The temperature dependence of the rate of formation of the surface layer is described by the equation

2. The rate of formation of the surface layer is described by the equation

3. The rate of formation of the surface layer is described by the equation

4. The rate of formation of the surface layer is described by the equation

5. The rate of formation of the surface layer is described by the equation

6. The rate of formation of the surface layer is described by the equation







BARANOVSKIY, V.Ye.; SHIMANSKIY, Yu.I.; GOLIK, A.Z.

Heat of evaporation of the ternary system ethyl alcohol-butyl alcohol - acetone. Ukr.khim.zhur. 28 no.4:484-486 '62.

(MIRA 15:8)

1. Kiyevskiy gosudarstvennyy universitet imeni T.G.Shevchenko.  
(Ethyl alcohol) (Butyl alcohol) (Acetone)  
(Heat of evaporation)

GOLIK, A.Z.; RYNDICH, N.A.; NUZHNYI, V.M.; GALAGAN, Yu.

Velocity of ultrasound and the compressibility of alcohol -  
acetone - water solutions. Ukr.khim.zhur. 28 no.4:506-510 '62.  
(MIRA 15:8)

1. Kiyevskiy gosudarstvennyy universitet imeni T.G.Shevchenko.  
(Alcohols) (Acetone) (Ultrasonic waves—Speed)

GOLIK, A.Z.; IVANOVA, I.I.

Molecular structure, density, compressibility, and shearing  
viscosity of n.paraffins in the liquid state. Zhur.fiz.khim. 36  
no.8:1768-1770 Ag '62. (MIRA 15:8)

1. Kiyevskiy gosudarstvennyy universitet.  
(Liquids) (Paraffins)

GOLIK, A.Z. [Holyk, O.Z.]; KUCHINKA, M.Yu. [Kuchynka, M.IU]

Temperature-time dependence of the strength of polymers at a  
constant tension rate. Ukr. fiz. zhur. 8 no.4:479-486 Ap '63.  
(MIRA 16:8)

1. Kiyevskiy gosudarstvennyy universitet im. Shevchenko,  
(Polymers--Testing)

BARANOVSKIY, V.I., COLIK, A.E.

Latent heat of vaporization of water-alcohol solutions, Ukr.  
khim. zhur. 29 no.2:137-141, '53. (MIRA 16:5)

1. Kiyevskiy gosudarstvennyy universitet im. T.G. Shevchenko.  
(Heat of evaporation) (Alcohols)

GOLIK, A.Z.; ADAMENKO, I.I.; SHOLPAN, P.F.

Effect of molecular interaction on the compressibility and  
viscosity of liquids. Ukr. fiz. zhur. 9 no.4:412-416 Ap '64.  
(MIRA 17:8)

1. Kiyevskiy gosudarstvennyy universitet.

"APPROVED FOR RELEASE: 09/24/2001

CIA-RDP86-00513R000515720010-0

1. The first part of the document is a list of the names of the individuals who were involved in the project. The names are listed in alphabetical order and are followed by their respective positions. The names are: [illegible]

APPROVED FOR RELEASE: 09/24/2001

CIA-RDP86-00513R000515720010-0"



During the past few years, the U.S. has been

concerned with the possibility of a

new arms race between the U.S. and the

U.S.S.R. in the field of space weapons.

The U.S. has been concerned with the possibility of a

new arms race between the U.S. and the

(1)  $\frac{1}{2} \leq \frac{1}{2} \leq \frac{1}{2}$

**APPROVED FOR RELEASE: 09/24/2001**

**CIA-RDP86-00513R000515720010-0"**

L 51442-65 EWT(m)/EPF(o) Pr-1 RM

ACCESSION NR: AP5011070

UR/0185/65/010/004/0443/0449

AUTHOR: Holyk, O. Z. (Golik, A. Z.); Adamenko, I. I.

TITLE: Compressibility and molecular structure of liquids. I. Compressibility of n-paraffins and of their mutual solutions

SOURCE: Ukrayins'kyi fizychnyy zhurnal, v. 10, no. 4, 1965, 443-445

TOPIC TAGS: n-paraffin, molecular structure, compressibility, liquid state, activation energy, viscous flow, intermolecular force

ABSTRACT: The authors investigate the compressibility of liquids having an identical molecular structure and the same type of intermolecular forces (the n-paraffins: n-heptane, n-octane, n-nonane, n-undecane and n-dodecane). It is shown that the compressibility polytherms of these liquids lie the lower the deeper the potential well on the molecular interaction curve and the larger the activation energy of viscous flow. It is also shown that under certain conditions it is possible to attain coincidence of the polytherms of compressibility of solutions of paraffins and pure substances or other solutions of paraffins of different composi-

Card 1/2

L 51442-65

ACCESSION NR: AP5011070

tions (these substances are called iso-compressible). Iso-compressible substances have identical activation energy of viscous flow and identical energy of intermolecular interaction. The adiabatic compressibility polytherms of the investigated normal paraffins are well described by the empirical formula  $\beta_{ad} = \beta_0 \exp \alpha T$  in which the constant  $\alpha$  is inversely proportional to the viscous-flow activation energy. The dependence of the adiabatic compressibility on the potential of the intermolecular interaction is in good agreement with modern statistical and model theories of liquids. Orig. art. has: 5 figures, 6 formulas, and 4 tables.

ASSOCIATION: Kyivskyy derzhuniversitytet im. T. G. Shevchenka [Kiyevskiy gosudarstvennyy universitet im. T. G. Shevchenko] (Kiev State University)

SUBMITTED: 21 Nov 64

ENCL: 00

SUB CODE: 00, NE

NR REF SCV: 003

OTHER: 003

*me*  
Card 2/2



ACC NR: AP7004553

SOURCE CODE: UR/0185/66/011/007/0797/9801

AUTHOR: Golik, A. Z.; Chelpan, P. P.; Tarasenko, O. V.

ORG: Kiev State University Im. T.H. Shevchenko (Kyyvs'kyy derzhunivorsitet)

TITLE: Velocity of ultrasonic vibrations and compressibility of liquid siloxanes

SOURCE: Ukrayins'kyy fizychnyy zhurnal, v. 11, no. 7, 1966, 797-801

TOPIC TAGS: siloxane, temperature dependence, ultrasonic vibration

ABSTRACT: The authors investigated the temperature dependence (within the range of 0 - 200°C) between the velocity of ultrasonic vibrations and the adiabatic compressibility of linear methyilsiloxanes - octamethyltrisiloxane, decamethyltetrasiloxane, dodecamethylpentasiloxane, cyclic methylsiloxanes - octamethylcyclotetrasiloxane, decamethylcyclopentasiloxane, and phenylmethylsiloxanes - heptamethylphenyltrisiloxane, pentamethyltriphenyltrisiloxane, octamethyldiphenyltetrasiloxane.

It is determined that the temperature dependence of ultrasonic velocity at high temperatures deviates from the linear dependence. The adiabatic compressibility obeys an exponential law over a small range of temperatures only. It is shown that the compressibility of siloxanes decreases with the increase of the intermolecular force potential and the co-ordination number.

Orig. art. has: 4 figures, 3 formulas and 2 tables. [JPRS: 37,333]

SUB CODE: 20,07 / SUBM DATE: 11Dec65 / ORIG REF: 009

Card 1/1

GOLIK, P.K., vrach

Simplified apparatus for the simultaneous injection of oxygen  
and solutions. Zd-ny.Kazakh. 17 no.9:48-50 '57.

(MIRA 12:6)

1. Iz oblastnogo kozhno-venerologicheskogo dispansera Severo-  
Kazakhstanskoy oblasti Kazakhstoy SSR.

(MEDICAL INSTRUMENTS AND APPARATUS) (INJECTIONS, HYPODERMIC)

65-38-6-4/43

AUTHOR: Golik, G., Senior Inspector-Pilot, DCSAAF Republic Committee of the Ukraine (Kyiv) (Respublikanskiy komitet DCSAAF Ukrayiny)

TITLE: Model-airplane Builders Prepared for Sports Combat (Aviatsionnyy gosportk sportivnyy boevoye)

PERIODICAL: Vyzhivaya molod, 1958, No. 6, p. 2 (USSR)

ABSTRACT: The author states that teams of some 270,000 model airplane builders are now being trained by 4,000 public instructors in DCSAAF primary organizations in the Ukraine. Personalities mentioned include: USSR champion Ye. Kondratenko, N. Dem'yachenko, Ye. Kucherenko, Ya. Usik, M. Olenkasskiy, V. Sheremet. In June and July (1958), 22,000 model airplane builders will compete at the Sportsman Games.

ASSOCIATION: DCSAAF Republic Committees of the Ukraine

1. Airplane-Model building

Card 1/1



GOLIK, G.

At gatherings. Kryl. rod. 14 no.2:18 F '63.

(MIRA 16:4)

1. Nachal'nik otdela aviatsionnoy podgotovki respublikanskogo komiteta Dobrovol'nogo obshchestva sodeystviya armii, aviatsii i flotu.

(Ukraine--Parachuting)

ACC NR: AD-1111

SOURCE CODE: UR/0079/05/036/09/1636/1639

AUTHOR: Shteyn, V. A.; Golik, L. A.; Likhner, B. Ya.; Derkach, G. I.

ORG: Institute of Organic Chemistry, Academy of Sciences, USSR (Institut organicheskoy khimii Akademii nauk USSR)

TITLE: Monoalkylamides of alkyl methylphosphonates

SOURCE: Zhurnal obshchey khimii, v. 36, no. 9, 1966, 1636-1639

TOPIC TAGS: insecticide, ~~monoalkylamine-alkyl methylphosphonate~~, ~~ORGANIC AMIDE~~, ~~PHOSPHONATE~~, ~~PHOSPHORIC ACID~~

ABSTRACT: In a search for new insecticides, a series of monoalkylamides of alkyl methylphosphonates was obtained by the reaction of methylphosphonic acid chloride with primary amines in the presence of triethylamine in an ether solution at room temperature:



Composition and properties of the amides are given in the table.

Card 1/4

UDC: 547.26'118

ACC NR: AP6031382

Table 1. Monoalkylamides of alkyl methylphosphonates

CH <sub>3</sub> P(O)(O <sub>2</sub> )NHR'					
R	R'	yield, %	bp (p, mm)	d <sub>4</sub> <sup>20</sup>	n <sub>D</sub> <sup>20</sup>
CH <sub>3</sub>	CH <sub>3</sub>	a, 37	72-73° (0.02)	1.1288	1.4423
CH <sub>3</sub>	C <sub>2</sub> H <sub>5</sub>	a, 58	78-79 (0.02)	1.0779	1.4402
CH <sub>3</sub>	н <sub>3</sub> о-C <sub>3</sub> H <sub>7</sub>	a, 42	81-83 (0.03)	1.0402	1.4573
CH <sub>3</sub>	н.-C <sub>4</sub> H <sub>9</sub>	a, 36	95-96 (0.1)	1.0192	1.4424
C <sub>2</sub> H <sub>5</sub>	CH <sub>3</sub>	6, 82 (69)	86-88 (0.5)	1.0885	1.4590
C <sub>2</sub> H <sub>5</sub>	C <sub>2</sub> H <sub>5</sub>	6, 72	91-93 (0.4)	1.0482	1.4372
C <sub>2</sub> H <sub>5</sub>	н <sub>3</sub> о-C <sub>3</sub> H <sub>7</sub>	6, 78 (62)	66-67 (0.03)	0.9975	1.4317
C <sub>2</sub> H <sub>5</sub>	н.-C <sub>4</sub> H <sub>9</sub> **	a, 54 (11)	100-101 (0.1)	0.9971	1.4479
iso-C <sub>3</sub> H <sub>7</sub>	CH <sub>3</sub>	6, 81 (55)	73-75 (0.06)	1.0372	1.4719
iso-C <sub>3</sub> H <sub>7</sub>	C <sub>2</sub> H <sub>5</sub>	6, 73	69-71 (0.03)	1.0403	1.4735
iso-C <sub>3</sub> H <sub>7</sub>	н <sub>3</sub> о-C <sub>3</sub> H <sub>7</sub>	6, 63	85-87 (0.07)	0.9863	1.4718
Card 2/4 iso-C <sub>3</sub> H <sub>7</sub>	н.-C <sub>4</sub> H <sub>9</sub>	a, 54 (13)	138-139 (11)	0.9712	1.4376

ACC NR: AP-1

Elemental Analysis (Calcd.)				
Found	Calc.	Element, %	Formula	Calculated
28.89	29.13	N 11.43	$C_5H_{15}NO_2P$	N 11.56
33.54	33.65	CH <sub>3</sub> O 22.53	$C_4H_{12}NO_2P$	CH <sub>3</sub> O 22.53
38.12	38.36	CH <sub>3</sub> O 20.65	$C_5H_{14}NO_2P$	CH <sub>3</sub> O 20.53
42.92	42.98	CH <sub>3</sub> O 18.74	$C_6H_{16}NO_2P$	CH <sub>3</sub> O 18.79
33.32	33.65	N 10.21	$C_4H_{12}NO_2P$	N 10.22
37.92	38.36	N 9.22; P 20.53	$C_5H_{14}NO_2P$	N 9.27; P 20.43
43.08	42.98	P 18.59	$C_6H_{16}NO_2P$	P 18.75
47.39	47.60	N 7.58	$C_7H_{13}NO_2P$	N 7.61
38.03	38.36	N 9.34	$C_5H_{14}NO_2P$	N 9.27
42.53	42.98	N 8.43	$C_6H_{13}NO_2P$	N 8.63
47.11	47.60	N 7.99; P 17.34	$C_7H_{13}NO_4P$	N 8.01; P 17.28
52.13	52.22	N 7.28; P 16.04	$C_8H_{20}NO_4P$	N 7.25; P 16.05

Card 3/4

ACC NR: AP6031382

These amides have strong insecticidal properties but are very toxic to domestic animals. Monoalkylamides of alkyl methylphosphonates react with tert-butyl hypochlorite to form N-chloro-N-alkylamides of alkyl methylphosphonates. The reaction takes place in chloroform at 20—30°C. [WA-50; CBE No. 12]

SUB CODE:06,07/ SUBM DATE: 17Jul65/ ORIG REF: 003/ OTH REF: 014/

Card 4/4

KHODCHENKO, L.P., inzhener; GOLIK, G.I., inzhener.

Standard metallic edge fittings for construction yards.

Shakht.stroi. no.4:25-27 Ap '57. (MLA 10:7)

(Building materials industry--Equipment and supplies)

GOLIK, G.Kh., student IV kursa; MEDVEDTS, I.P., student V kursa

Professor Petr Ivanovich Shatilov, founder of the original  
Russian school of therapeutics. Klin.med. zh. no.8:47-91  
Ag '56. (MIRA 12:8)

1. Iz kafedry propedavтики vnutrennikh bolezney (zav. -  
zasluzhennyy iatel' nauki prof. V.M.Kozan- Yasnyy) lechebnogo  
fakul'teta Khur'kovskogo meditsinskogo instituta i 26-v klini-  
cheskoy bol'nitsy (glavnyy vrach M.M.Gorodnichenko).

(BIOGRAPHIES

Shatilov, Petr I.)





**CIA-RDP86-00513R000515720010-0"**

1. The first part of the document is a list of the names of the individuals who were involved in the project. The names are listed in alphabetical order.

2. The second part of the document is a list of the dates when the individuals were involved in the project. The dates are listed in chronological order.

3. The third part of the document is a list of the locations where the individuals were involved in the project. The locations are listed in alphabetical order.

4. The fourth part of the document is a list of the activities that the individuals were involved in. The activities are listed in alphabetical order.

5. The fifth part of the document is a list of the results of the project. The results are listed in alphabetical order.

WILEK, K.N. [Holvik, K.M.]

Diurnal and seasonal dynamics of the intensity of photosynthesis in the sweet cherry, cherry, plum and apricot. Ussr. bot. zhur. 19 no.3: 20-27 '6 . (MIRA 15:7)

1. Institut botaniki AN USSR, otdel fotosintaza.  
(Photosynthesis) (Fruit trees)

GELIK, V.M. [hol?], 1961

Intensity of photosynthesis and transpiration in the leaves  
of different sides of the crown in the sweet cherry, sour  
cherry, plum and apricot depending on the light. Her. Bot.  
Akad. Nauk SSSR 1961. (1961) 1961

1. Intensity of photosynthesis and transpiration in the leaves  
of different sides of the crown in the sweet cherry, sour  
cherry, plum and apricot depending on the light.

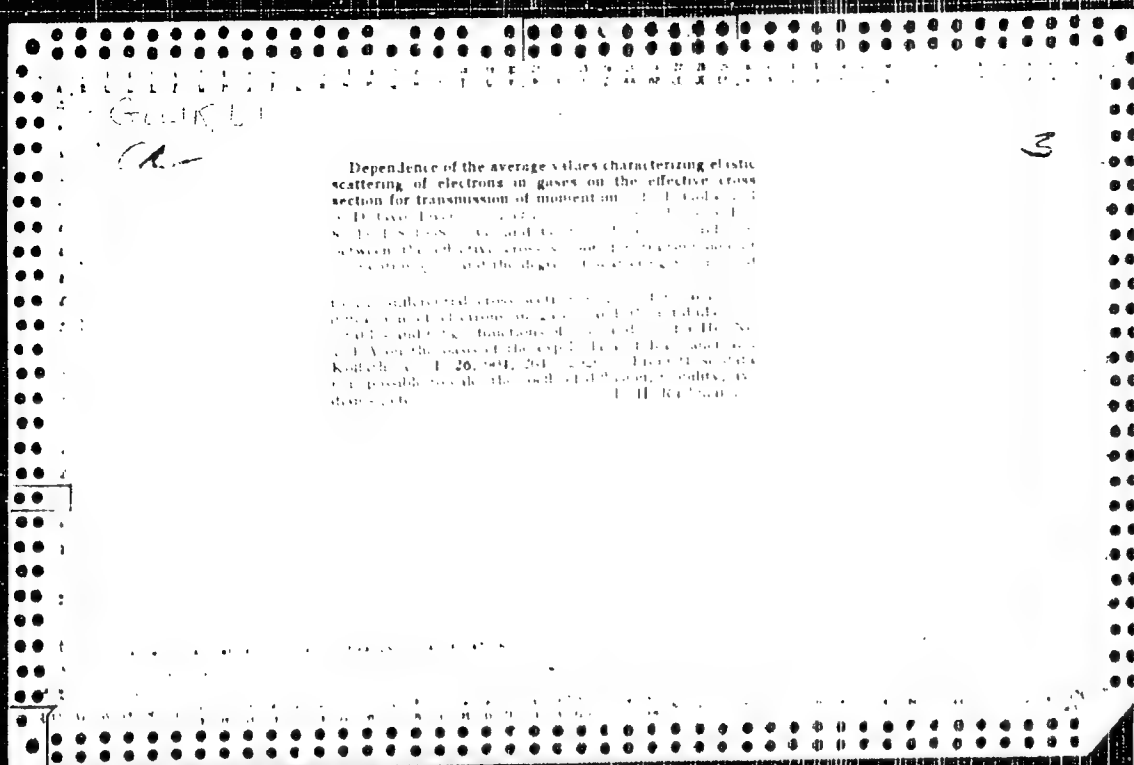
GOLIK, K.N. [Halyk, K.M.]

Effect of shade on the intensity of photosynthesis and  
transpiration in the representatives of Prunaceae. Ukr. bot.  
zhur. 71 no. 1: 18-24 '64. (SCLBA 1743)

1. Otdel fotosinteza Instituta botaniki AN UkrSSR.









USSR/Physics  
Electron Microscope  
Arce, Mercury

Jul 49

"Electronic Optical Phenomena in the Plasma of a  
Mercury Arc Under Low Pressure," L. I. Golik, S. V.  
Spirak, Phys Faculty, Moscow State U Imeri M. V.  
Lomonosov, 9 pp

"Zhur Tokh Fiz" Vol XIX, No 7

Shows that in the field of a magnetic lens acting  
on a stationary plasma, contractions of two types  
develop: (1) diffusive-plasmaic, and (2)  
electronic optic. Second case develops for a  
rapid change in potential and an artificial con-  
traction in plasma. Measured distribution in-  
cludes along radius of ray and clarified in-  
fluence of second magnetic lens influences focusing;  
Magnitude of discharge current influences field in  
as it does all effects of contraction with increase  
the plasma. Decrease in contraction in proportion of  
in current is due to increase in electrons in com-  
parison with direction. Submitted 29 Jun 48.

51/49759

GOLIK, I. I.

USSR/Physics - Plasma

Oct 53

"Electron Optical Phenomena in Focussing and Stationary Plasma in Mercury Vapors," I.I. Golik (deceased) and G.V. Spivak, Chair of Electron Optics Vest Mos Univ, Ser Fizikomat i Yest Nauk, No 7, pp 117-123

In their previous works (see Zhur Ekst Teor Fiz, Vol 23, 1953) at their lab the authors established the presence of the phenomena of plasma focussing, which occurs under the action of external and internal electrical and magnetic fields. Their purpose here is to study the phenomena of contraction

273T97

(necking) of stationary plasma toward the axis of symmetry in a strong and concentrated external magnetic field.

GALIK, L. I., MORAVENSKIY, I. I., and SHEVCHUK-BERS, E. I.

"Application of a Differential Thermo-couple for the  
Investigation of Mass Transfer at Drying Cellulose  
Materials."

Report submitted for the Conference on Heat and Mass Transfer,  
Minsk, USSR, June 1961.



135

S. C. S. 10-11-68 0-3023  
S. C. S.

... ..

ASSOCIATION: The Association of American Universities (AAU) is a non-profit organization that represents the interests of the nation's leading research universities. It was founded in 1900 and has since grown to include 50 member institutions across the United States. The AAU's primary focus is on promoting research, scholarship, and the advancement of higher education. It also works to address issues related to the role of universities in society, such as access, equity, and the environment. The AAU is a key player in the higher education landscape, and its actions and policies have a significant impact on the academic community.

[illegible]

2000年12月29日  
 2001年1月1日

SOLOV, A., "On the mechanical properties of polymers," in 1.

to indicate more precisely the mechanical properties of polymers.  
Makulev, prof. 1961 no. 12:15-16 (1961). (MIRA 1961)

1. V. G. Goryunov, "On the mechanical properties of polymers," in 1.

GOLIK, M. G.

"Scientific Principles Underlying the Storage of Ear and Shelled Corn and Their Practical Application." Dr Agr Sci, Khar'kov Agricultural Inst, Moscow, 1953. (RZhSiol, No 6, Nov 54)

Survey of Scientific and Technical Dissertations Defended at USSR Higher Educational Institutions (11)

SO: Sum. No.521, 2 Jun 55

Chemical Abst.

Vol. 48 No. 8

Apr. 25, 1954

Fats, Fatty Oils, Waxes, and

Detergents

A method is described for the determination of the content of linolenic acid in rapeseed oil. The method involves the use of a special reagent which reacts with the linolenic acid to form a colored compound. The color intensity is measured and compared with a standard curve to determine the content of linolenic acid. The method is simple and accurate and can be used for the determination of linolenic acid in other oils and fats. Vladimir N. Kuznetsov



GOLIK M.G.

VORONTSOV, O.S.; GOLIK, M.G.; DELIDOVICH, V.N.; KLEYEV, I.A.; KOZ'-  
MINA, N.P., doktor biologicheskikh nauk, professor; SOSEDOV, S.I.  
FESTA, N.Ya.; CHUKHAR'KO, Z.T.; GIL'MAN, D.Ya., redaktor; LA-  
BUS, G.A., tekhnicheskii redaktor.

[Grain storage; management and equipment] Organizatsiia i tekhnika  
khraneniia zerna. Moskva, Izd-vo tekhn. i ekonomicheskoi lit-ry.  
1954. 358 p. [Microfilm] (MLBA 7:10)  
(Grain--Storage)

Golik, M.G.

U.S.S.R.

✓ Accumulation of fat and its distribution in the corn grain.  
M. G. Golik. *Problemy Zerna, Akad. Nauk S.S.S.R.*  
~~1954~~ (1954). --In the course of vegetation, corn  
grain contains much fat, reaching 1.87% at ripeness. Em-  
bryo tissues contain more fat than do other parts of the  
grain. The embryo also bears the greater proportion of  
enzyme systems, which are of importance in storability of  
corn. Thus the state of the embryo is more important as a  
storability factor in corn than in other grain. G. M. K...

BAUM, A., kandidat tekhnicheskikh nauk; GOLIK, M., kandidat sel'sko-khozyaystvennykh nauk.

Transference of moisture in stored grain. Muk.-elev.prom. 20  
no.3:3-6 Mr '54. (MLRA 7:7)

1. Vsesoyuznyy nauchno-issledovatel'skiy institut zerna i  
produktov ego pererabotki.  
(Grain-Storage)

GOLIK, M. G.

Changes in the enzymic activity of ripening corn seeds.  
M. G. Golik. *Trudy Vsesoyuz. Nauch. Issledovatel. Inst.*  
*Zerna* 1954, No. 27, 66-72; *Referat Zhur. Khim., Biot.*  
*Khim.* 1954, No. 11792. The activity of dehydrases,  $\alpha$ -  
and  $\beta$ -amylase, and of lipase in the corn seeds during the  
period of milk formation is high. As the seeds ripen the  
activity of such enzymes gradually recedes until at the stage  
of complete ripening enzymic activity is not detected. It is  
suggested that the possible persistence of amylase may be  
one of the factors responsible for the easy spoilage of corn  
harvested before its full maturity. B. S. Levin.

GOLIK, M. G.

Golik, M. G.: Fiziologo-biokhimiicheskie osnovy khraneniya zhirnykh (Physiological-Biochemical Bases for the Storage of Corn). Moscow: Izdatel. Akad. Nauk S.S.S.R. 1955. 223 pp.

GOLIK, M., doktor sel'skokhozyaystvennykh nauk.

Particular aspects of corn storage. Muk.-elev.prom. 21 no.10:4-7  
0 '55. (MLBA 9:1)

1.Vysshaya zagotovitel'naya shkola.  
(Corn (Maize)--Storage)

GOLIK, Mikhail Grigor'evich, doktor sel'skokhozyaystvennykh nauk, professor;  
KRETOVICH, V.L., professor, doktor biologicheskikh nauk, redaktor;  
GEL'MAN, D.Ya., redaktor; GOLUBKOVA, L.A., tekhnicheskiy redaktor

[Storage of corn: scientific principles] Khraneniye kukuruzy; nauchnye osnovy. Pod red. V.L.Kretovicha. Moskva, Izd-vo tekhn. i ekon. lit-ry po voprosam mukomol'no-krupianoj, kombikormovoj promysl. i elevatorno-skadskogo khozjaistva. Khleboizdat, 1956. 115 p.  
(Corn (Maize)--Storage) (MLRA 10:3)

GOLIK, M. G.

✓ Dehydrogenases of developing grain of maize. M. G. Golik. *Bizhnim. Zerna*, Sverdlovsk 1936, No. 3, 166-67. During maturation of maize grain there is a gradual decrease in the rate of respiration of the grain. Along with this decrease in rate of respiration there was observed a decrease in the activity of dehydrogenases, indicating that the lowering of the respiratory gas metabolism is connected with the lowering of the activity of the oxidation-reduction enzymic systems, mainly of dehydrogenases.



GOLIK, M., doktor sel'skokhozyaystvennykh nauk, prof.; RYAZANTSEV, P.

Effectiveness of drying ear seed corn with hot air in Krasnodar Territory. Mik.-elev. prom. 24 no.8:14-16 Ag '58. (MIRA 11:10)

1. Nachal'nik Kraenodarskogo upravleniya khleboproduktov (for Ryazantsev).

(Kraenodar Territory--Corn (Maize)--Drying)

GOLIK, Mikhail Grigor'yevich, doktor sel'khoz. nauk, prof. Prinsipal ucha-  
stiye KREYMERMAN, G.I., kand. tekhn. nauk; KRETOVICH, V.I., doktor  
biolog. nauk, prof., otv. red.; GLAZUNOV, Ye.A., red. izd-va;  
SHAFFANSKAYA, M.Z., red. izd-va; POLYAKOVA, T.V., tekhn. red.

[Scientific principles in the storage and processing of corn]  
Nauchnye osnovy khraneniia i obrabotki kukuruzy. Moskva, Izd-vo  
Akad. nauk SSSR, 1961. 347 p. (MIRA 14:10)  
(Corn (Maize))--Storage)

GOLIK, M., doktor sel'skokhozyaystvennykh nauk; FOMIN, N., kand. tekhn. nauk

Evaluating the germinative capacity of corn. Dokl.-elev. prom.  
27 no.8:25-27 Ag '61. (MIRA 14:7)

1. Vsesoyuznyy zaochnyy institut pishchevoy promyshlennosti.  
(Corn (Maize)  
(Germination)

GOLIK, M., professor-doktor

"Provisional recommendations on the receiving, disposing, and storing of pulse crop seeds (peas, forage beans, and lupine)."  
Reviewed by M. Golik. Muk.-elev. prom. 28 no.10:32 and p. 3 of cover 0 '62. (MIRA 10:1)

1. Vsesoyuznyy zaochnyy institut pishchevoy promyshlennosti.  
(Legumes) (Seeds--Storage)